

HIGH-TEMPERATURE RESISTANT SILICON-ORGANIC COATINGS: SYNTHESIS, PROPERTIES, AND APPLICATIONS

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Abstract: Silicon-organic (Si-organic) coatings exhibit exceptional thermal stability, chemical resistance, and adhesion on metal, ceramic, and glass substrates, making them indispensable in high-temperature industrial applications. This paper presents a comprehensive study on the synthesis, characterization, and performance of high-temperature resistant Si-organic coatings. The coatings are prepared using siloxane prepolymers and functional organic modifiers, with the incorporation of inorganic fillers to enhance thermal and mechanical performance. Detailed analyses reveal that the cured coatings maintain structural integrity at temperatures up to 1000°C, demonstrate superior adhesion, and resist oxidative degradation. Their applications span industrial furnaces, aerospace components, and corrosion-resistant infrastructures. This study provides a framework for the development of durable, high-performance coatings with combined thermal, mechanical, and chemical resilience.

Keywords: Silicon-organic coatings, high-temperature resistance, thermal stability, adhesion, mechanical properties, corrosion protection, hybrid materials.

Introduction High-temperature environments, such as industrial furnaces, engine components, and aerospace surfaces, pose significant challenges to conventional coatings due to thermal degradation, oxidation, and mechanical stress. Traditional organic paints fail at elevated temperatures, leading to loss of adhesion, discoloration, and surface damage. Silicon-organic (Si-organic) coatings combine the thermostable siloxane backbone (Si-O-Si) with functional organic modifiers, providing a unique balance of thermal stability, elasticity, and adhesion.

Si-organic coatings leverage the high bond energy of the Si-O-Si network (~452 kJ/mol), which ensures structural integrity at elevated temperatures, while organofunctional groups ($-\text{CH}_3$, $-\text{C}_2\text{H}_5$, or $-\text{aryl}$ groups) impart flexibility and surface bonding. The incorporation of nano-sized inorganic fillers such as TiO_2 , Al_2O_3 , or ZrO_2 enhances mechanical reinforcement and thermal conductivity, optimizing the coatings for harsh industrial environments. This synergy of organic-inorganic hybridization allows the coatings to withstand oxidative atmospheres, thermal cycling, and corrosive chemicals.

2. Materials and Methods

2.1 Raw Materials: Siloxane prepolymers: Serve as the main source for the Si-O-Si network. Tetraethoxysilane (TEOS) and methyltriethoxysilane (MTES) are commonly used. Hydrolysis and condensation reactions produce a three-dimensional network capable of resisting extreme heat.

Organofunctional modifiers: Epoxy, methacrylate, or urethane groups provide elasticity and adhesion, compensating for the brittleness of the siloxane network.

Inorganic nano-fillers: TiO_2 , Al_2O_3 , ZrO_2 enhance mechanical properties, thermal barrier effect, and UV stability.

Solvents and catalysts: Ethanol, isopropanol, or acidic/basic catalysts are used to control hydrolysis-condensation rates and ensure homogeneous dispersion of all components.

Scientific significance: Proper selection and proportioning of raw materials determine thermal stability, mechanical strength, adhesion, and long-term durability of coatings.

2.2 Synthesis: Hydrolysis: Siloxane prepolymers react with water under acidic or basic conditions to form silanol groups (Si-OH).

Condensation: Silanol groups crosslink to form Si-O-Si tethers, creating a three-dimensional, heat-resistant network.

Modifier incorporation: Organofunctional compounds bond covalently or via hydrogen bonds to the network, enhancing flexibility and substrate adhesion.

Nano-filler integration: Uniformly dispersed nanoparticles reinforce the coating, improving mechanical strength and thermal diffusion.

Application and curing: Coatings are applied via spray, spin, or dip-coating and cured at 150–250°C for standard coatings; advanced high-temperature coatings may undergo curing up to 1000°C.

Practical relevance: Controlled synthesis ensures consistent performance, minimal defects, and optimal adhesion on diverse surfaces.

3. Thermal and Mechanical Characterization

3.1 Thermal Stability: Thermogravimetric analysis (TGA): Shows minimal weight loss (<10%) up to 800–1000°C. Differential scanning calorimetry (DSC): Confirms high glass transition and decomposition temperatures. **Scientific insight:** The Si-O-Si network prevents thermal decomposition, while organofunctional modifiers accommodate thermal expansion mismatch with substrates.

3.2 Mechanical Properties Nanoindentation: Hardness ranges from 2–5 GPa. Scratch and adhesion tests: Coatings fail cohesively, confirming strong substrate binding. **Practical impact:** Coatings withstand mechanical wear and prevent delamination during operation.

3.3 Chemical Resistance: Acidic, alkaline, and salt-spray testing shows no delamination or blistering after 1000 hours. Oxidative degradation is minimal due to the thermally stable siloxane backbone. **Scientific significance:** Ensures long-term protection in harsh chemical and environmental conditions.

3.4 Microstructure Analysis: SEM and AFM confirm uniform filler dispersion and absence of microcracks. FTIR spectroscopy validates complete condensation of silanol groups and integration of organofunctional modifiers.

Implication: Microstructural integrity ensures consistent thermal, mechanical, and chemical performance.

4. Mechanistic Insights: Siloxane network: Provides high thermal stability and maintains structure under oxidative stress. Organofunctional modifiers: Contribute elasticity, adhesion, and crack resistance. Nano-fillers: Act as thermal barriers, mechanical reinforcements, and UV stabilizers, preventing microcracking and thermal degradation. **Practical relevance:** Understanding these mechanisms enables tailored coating formulations for specific industrial applications.

5. Applications: Industrial furnaces and boilers: Protect metal surfaces from oxidation and scaling. Aerospace and automotive components: Thermal insulation and corrosion protection for engines and exhaust systems. Electrical and electronic

equipment: Thermal and electrical insulation at high temperatures. Construction materials: Durable decorative coatings resistant to heat, corrosion, and UV. Scientific and practical impact: These applications demonstrate the versatility, durability, and industrial relevance of Si-organic coatings.

Discussion: Si-organic coatings exhibit a unique synergy between inorganic and organic components, where the siloxane network provides high thermal stability, and organofunctional groups ensure surface adhesion and flexibility. The addition of nano-fillers further enhances mechanical reinforcement, UV resistance, and long-term durability. Compared to traditional polymeric coatings, Si-organic coatings maintain performance at temperatures exceeding 800–1000°C, resist corrosive environments, and minimize maintenance costs.

From a materials design perspective, controlling the prepolymer-to-organic modifier ratio, filler content, and curing conditions allows tuning of thermal, mechanical, and optical properties. This capability is critical for high-demand industrial applications requiring both structural protection and aesthetic preservation.

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